



# Photoinductive degradation of two estrogens by natural dissolved organic matter under simulated sunlight

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# PHOTOINDUCTIVE DEGRADATION OF TWO ESTROGENS BY NATURAL DISSOLVED ORGANIC MATTER UNDER SIMULATED SUNLIGHT

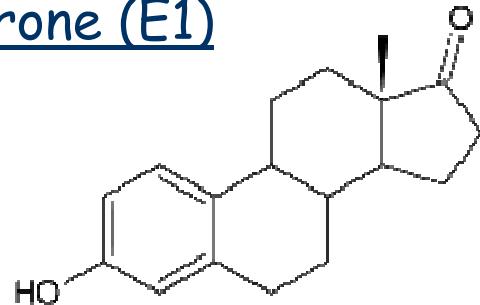
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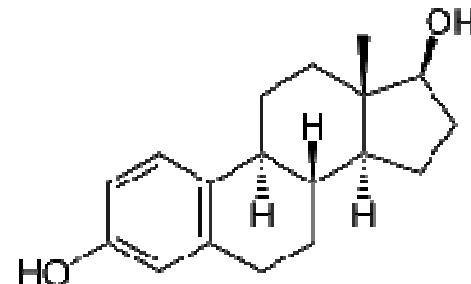
EMEC 10 Limoges 12/04/09

# Introduction

## Estrone (E1)



## 17 $\beta$ -estradiol (E2)



### ➤ Presence in the environment

- natural estrogens (steroidal hormones) present in aquatic media (through animals excretion)
- released into surface and ground waters through wastewater effluents
- detected in soils adjacent to agricultural fields fertilized with animal waste
- **E1 and E2 = two main estrogens detected in fresh and marine waters ~ in ng/L, Hohenblum 2004, Zuo 2006)**

# Introduction

## ➤ Effects on aquatic organisms

- impact on reproductive system and development of reproductive organs (Jobling 2002, Rodgers-Gray 2000)
  - **estrogenic effects** detected for vertebrates and invertebrates, fishes, amphibians, reptiles (feminisation, ... Orlando 2004, Brion 2004, Lai 2002)

## ➤ Human impacts

- **increased** incidents of breast, testicular and prostate **cancer**
- reduced fertility of men (Harrison 1997, Colburn 1995, Carlsen 1995)
- no conclusive relationships established between endocrine disruptor exposure and human health

## ➤ Photodegradation : half-lives under simulated solar system

- E1 : 4,7 h (Lin et Reinhard 2005)
- E2 : 13,6 h (Leech 2008)

# Introduction

## Natural Dissolved Organic Matter (DOM)

### ➤ Presence in the environment

- complex matrix
- natural decomposition of ecosystems : terrestrial and aquatic origins
  - composition and concentration depending on the nature and origin of the media (Thurman 1985)

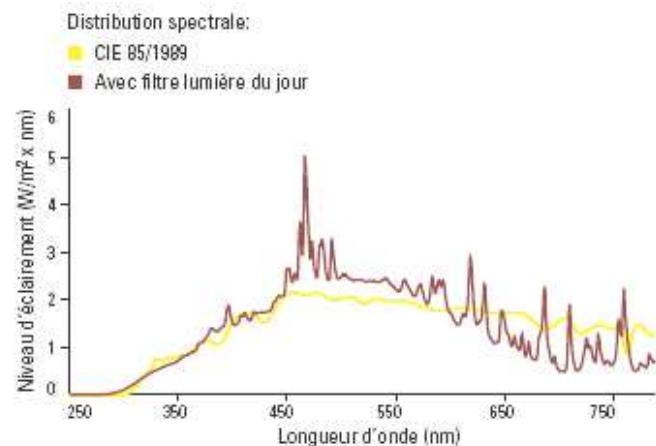
### ➤ Photodegradation

- degradation (break of aromatic structures Carvalho 2008)
- light absorption and **production of reactive species** ( $^1\text{O}_2$ ,  $\text{OH}^\circ$ ,  $\text{RO}^\circ$ , solvated electrons, ... Aguer 1999)

# Materials and Methods

## Photolysis

- Photodegradation under **simulated sunlight** (Suntest Atlas CPS+)



Suntest 8h, 250 W/m<sup>2</sup>, 900 kJ/h :  
- ~ 800 nM E1 or E2  
- 20 mg/L DOM  
- pH = 7

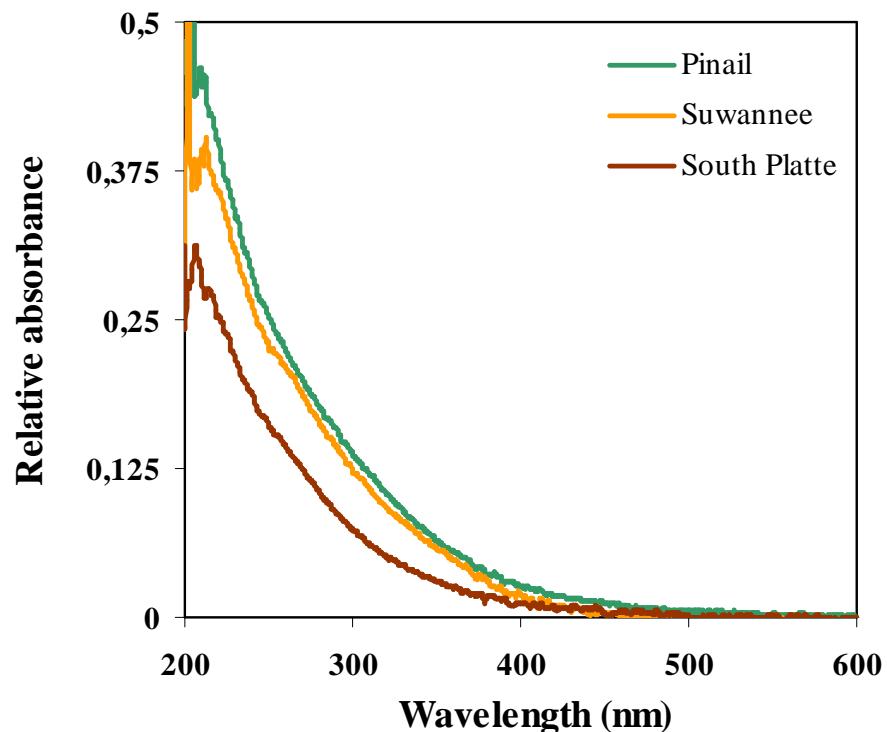
## Analysis

- HPLC-UV
- Spectrophotometer, Fluorimeter, TOCmeter

## Materials and Methods

### Optical properties of DOM

- 3 extracts used in photolysis experiments : fulvic acids of Pinail (France), Suwannee and South Platte (US rivers) at 20 mg DOM/L or 10 mg C/L



Order of UV-Visible absorbance

Pinail >

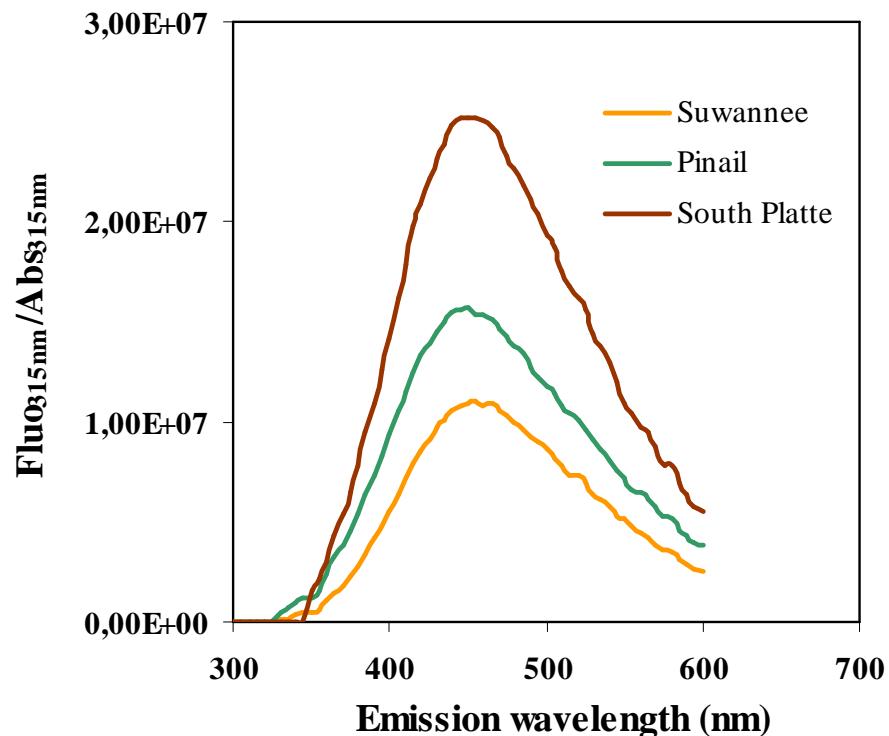
Suwannee >>

South Platte

## Materials and Methods

### Optical properties of DOM

- 3 extracts used in photolysis experiments : fulvic acids of Pinail (France), Suwannee and South Platte (US rivers) at 20 mg DOM/L or 10 mg C/L



Fluorescence/Absorbance Order

South Platte >  
Pinail >  
Suwannee



South Platte  
The most efficient

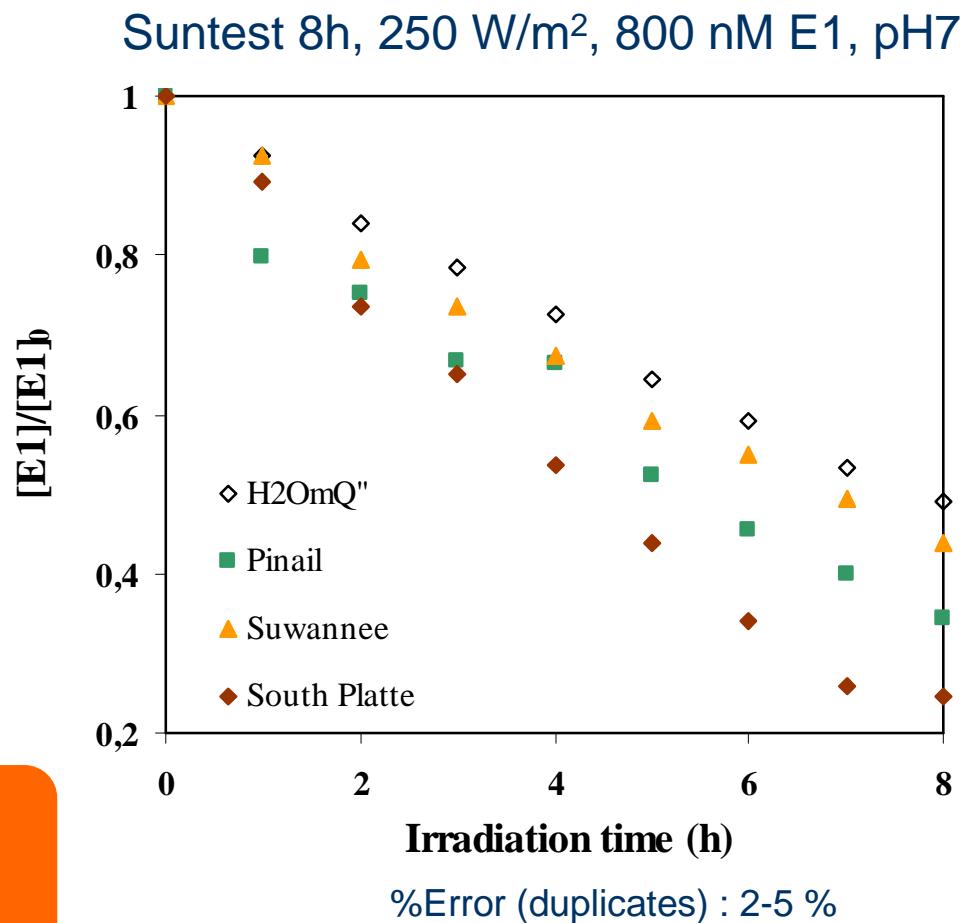
# Indirect photolysis

Nature of DOM Influence → E1

Photoinductive efficiency

South P. 76% >  
Pinail 66% >  
Suwannee 56%

Relationship between  
DOM nature –  
photoinductive efficiency



# Indirect photolysis

Nature of DOM Influence → E2

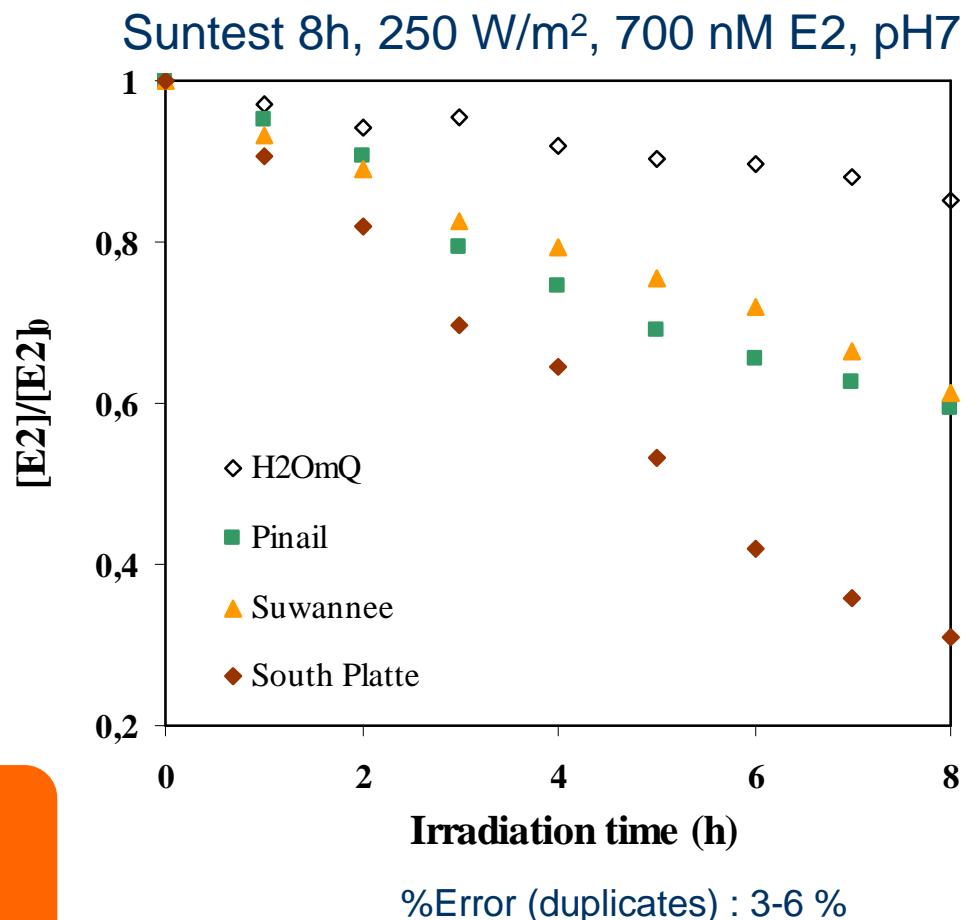
Photoinductive efficiency

South P. 69% >>  
Pinail 41% >  
Suwannee 39%

Relationship between  
DOM nature –  
photoinductive efficiency

The more fluorescent the  
more photoinductive efficient

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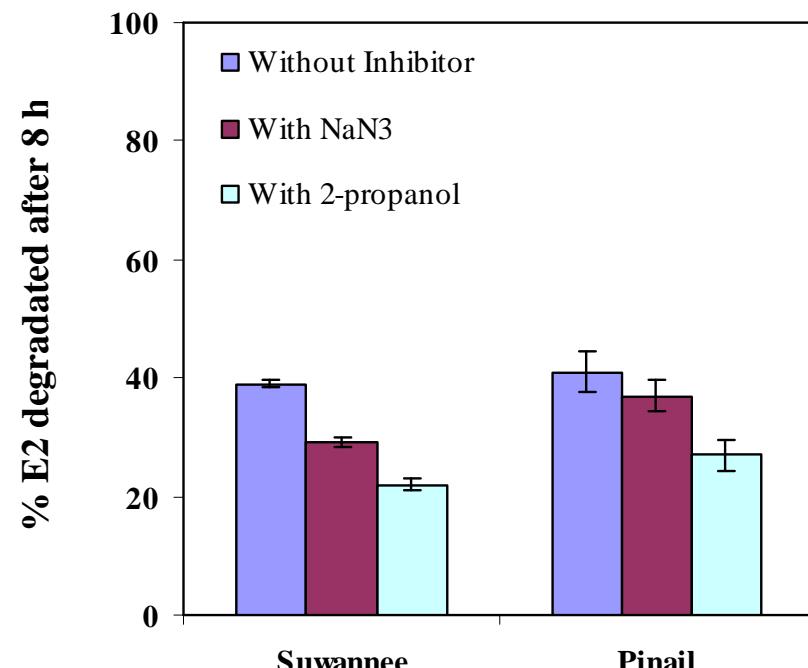
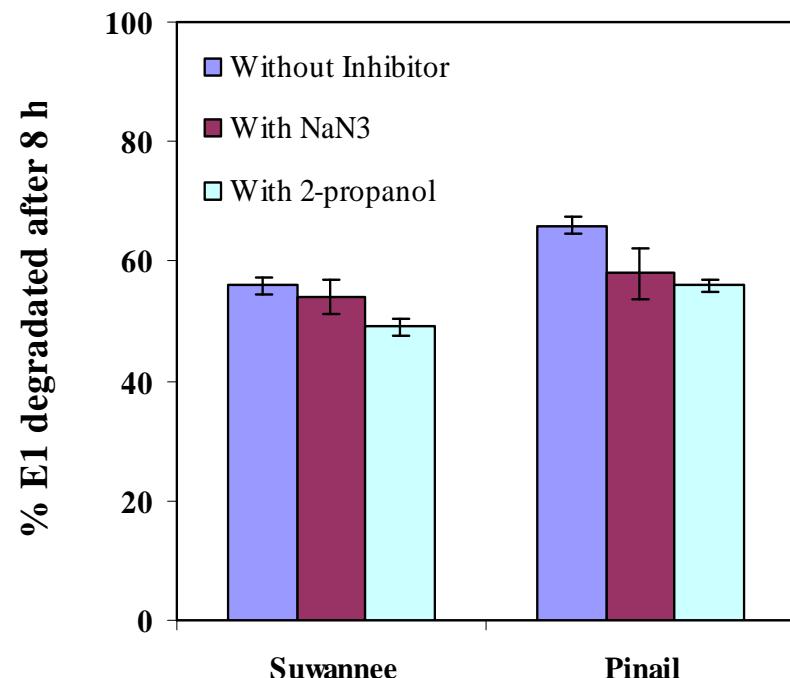


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# Indirect photolysis

## Reactive species

➤ Addition of inhibitors :  $\text{NaN}_3$  and 2-propanol



Photodegradation decrease with the addition of inhibitors

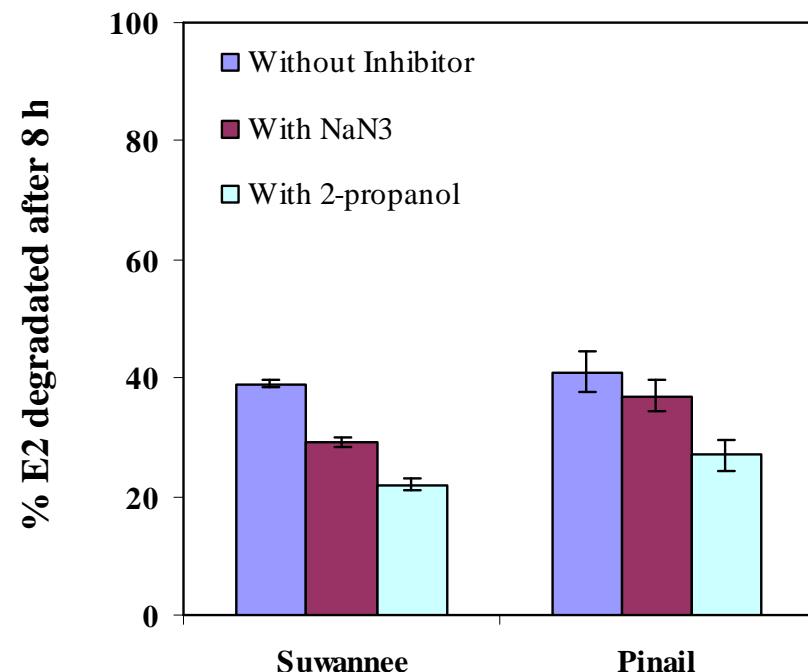
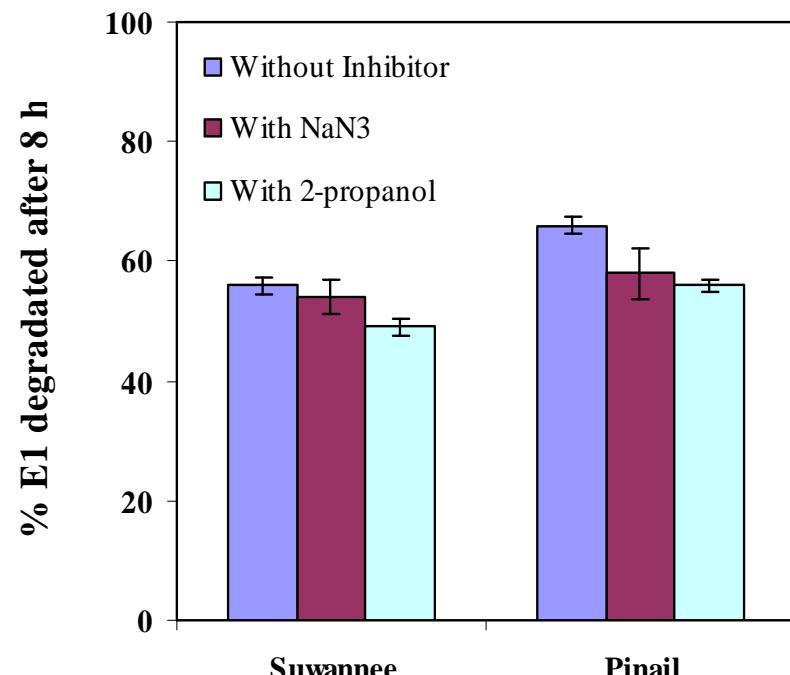
For every DOM :

Degradation without inhibitor > with  $\text{NaN}_3$  > with 2-propanol

# Indirect photolysis

## Reactive species

- Addition of inhibitors :  $\text{NaN}_3$  and 2-propanol



Reactive species in the inductive photodegradation

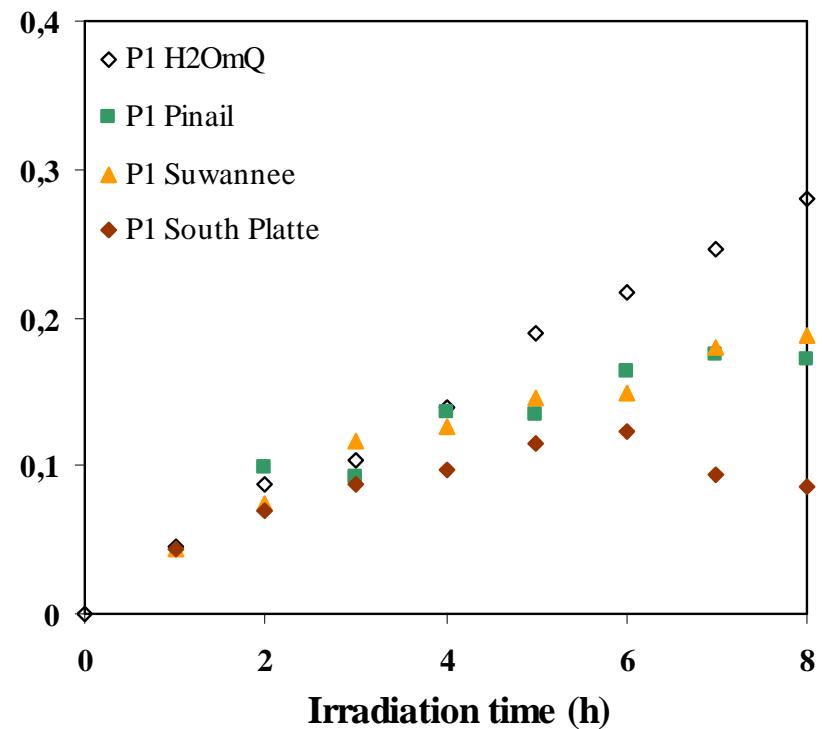
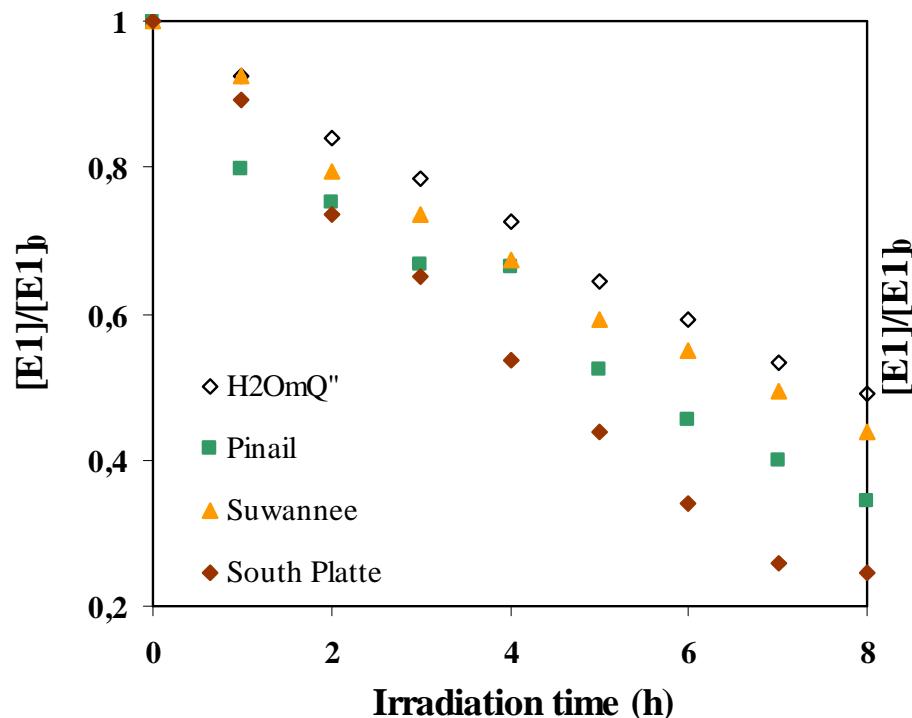
$^1\text{O}_2$  : participation in the photodegradation about 2 – 10 %

$\text{OH}^\bullet$  : participation in the photodegradation about 7 – 17 %

# Indirect photolysis

## Photoproduct (P1)

### ➤ Formation kinetics



# Indirect photolysis

## Photoproduct (P1)

### ➤ Formation kinetics

P1 formation decreases in the presence of DOM

Degradation of P1 by S. P.  
after 6 h

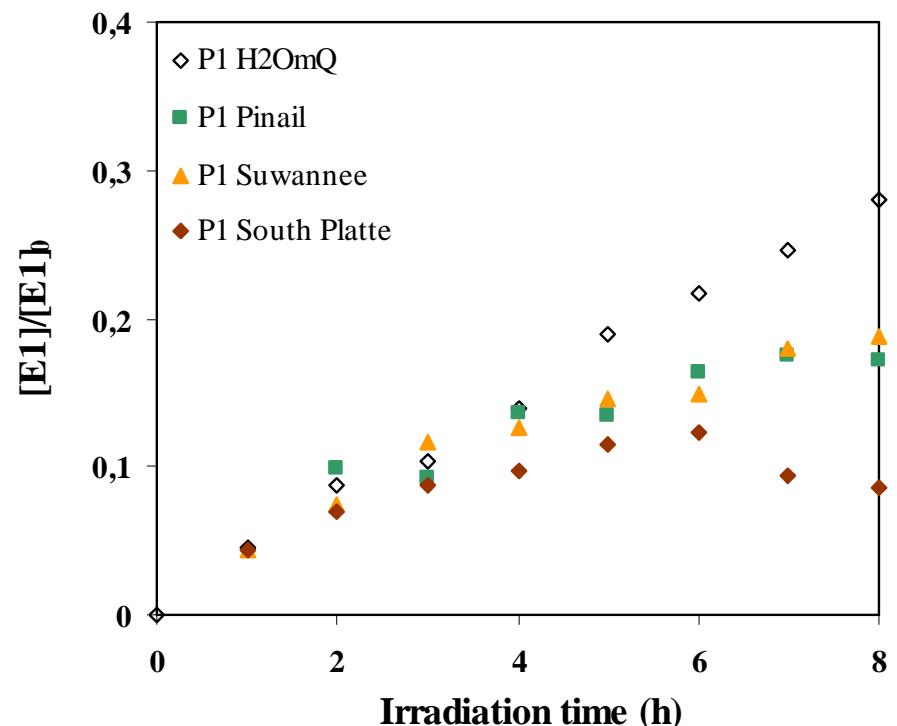


P1 degraded by DOM

or

Optical filter effect of DOM

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# Indirect photolysis

## Kinetics

- Pseudo-first order kinetic rate constants obtained for E1 and E2 during inductive photodegradation by DOM solutions

E1	$k_{O_2}$ (h <sup>-1</sup> )	$k_{O_2+NaN_3}$ (h <sup>-1</sup> )	$k_{O_2+Pro}$ (h <sup>-1</sup> )
H <sub>2</sub> OmQ	0,09	R <sup>2</sup> =0,998	
Pinail	0,13	R <sup>2</sup> =0,977	0,11 R <sup>2</sup> =0,994
Suwannee	0,10	R <sup>2</sup> =0,997	0,09 R <sup>2</sup> =0,964
South Platte	0,18	R <sup>2</sup> =0,981	* * 0,16 R <sup>2</sup> =0,978

E2	$k_{O_2}$ (h <sup>-1</sup> )	$k_{O_2+NaN_3}$ (h <sup>-1</sup> )	$k_{O_2+Pro}$ (h <sup>-1</sup> )
H <sub>2</sub> OmQ	0,02	R <sup>2</sup> =0,958	
Pinail	0,07	R <sup>2</sup> =0,985	0,05 R <sup>2</sup> =0,922
Suwannee	0,06	R <sup>2</sup> =0,993	0,04 R <sup>2</sup> =0,990
South Platte	0,14	R <sup>2</sup> =0,973	* * 0,11 R <sup>2</sup> =0,937

\* Not done

## Conclusion

- Photodegradation of E1 and E2 is possible under natural sunlight
- Observation of DOM photosensitivity
- Participation of singlet oxygen and hydroxyl radicals in the reaction
- Formation of a by-product

## Future works

- Identification of the by-product (LC-MS)
  
- Development on the relationship between nature of DOM and its photoinductive properties (RMN  $^{13}\text{C}$ , oxidation of DOM to the hydrophilic fraction)
  
- Study of by-product toxicity (ysters)

Thank you